

SYNERGY CHEMICAL INDUSTRIES

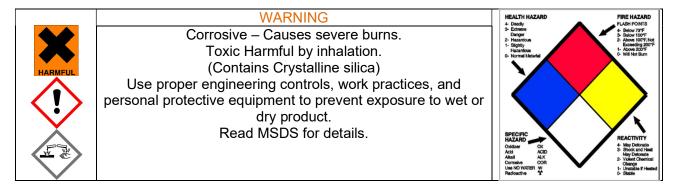
Material Safety Data Sheet

SECTION 1 – PRODUCT & MANUFACTURER INFORMATION

Product Identity: High Calcium, Hydrated Lime, Slaked Lime, Lime Hydrated Chemical Formula - Ca(OH)₂ Chemical Name - Calcium Hydroxide Ca(OH)₂ Chemical Family - Alkaline Earth Hydroxide

Manufacturer's Details: Name: Synergy Chemical Industries Factory Address: Kh.No.1110, Vil. Bhawanda,Teh. Khinvsar, Dist Nagaur, Pincode 341025 Emergency Contact No. 9833831479

Product use/s: Acid neutralizer, water treatment, PH correction, detergent additive, soil stabilization agent, asphalt, paints and concrete products, paper coating disinfectant etc.



SECTION 2 – COMPOSITION & INFORMATION ON CONTENTS

Grades of Hydrated Lime		Grade SP	Grade AA	Grade A	Grade B	Grade C	Grade D
Specifications	Min / Max	In %	In %	In %	In %	In %	In %
Available Lime as Ca(OH) ₂ %	Min	94-96	92	90	85	80	70
Active CaO %	Min	68-69	70	68	64	61	53
% Acid Insoluble	Max	0.5	1	1	2	4	8
% Moisture	Max	0.5	0.5	1	2	3	3
% Iron & Alumina (As Fe)	Max	Nil	In Traces	In Traces	In Traces	In Traces	In Traces
% Magnesia as MgO	Max	2	2	2	3	4	4
Mesh Size 100% Passing	Min	300	300	250	150	100	100

To convert the available lime content from CaO to Ca(oH)2 multiply the result by 1.32

SECTION 3 – HAZARDS IDENTIFICATION

Emergency Overview: Hydrated Lime is a white to off-white powder and odorless. It is corrosive to eyes and skin. Rules of entry in human body are as below:

- I) Skin Contact
- II) Ingestion
- III) Inhalation
- IV) Eye contact

Care should be taken to limit the exposure and prevent direct contact with eyes. Due to mishandling dust created may cause severe respiratory irritation. Further presence of quartz in the material in traces may be carcinogen. However as stated above sand in this product is minimal.

SECTION 4 – FIRST AID MEASURES

- 1) **Skin Contact**: Remove dry lime from body and wash large amount of water. May apply edible oil to smooth skin. If improvement not seen may call for medical help.
- 2) **Ingestion**: If patient is conscious small portion of glucose with water can be given. Do not induce vomiting. For unconscious person get immediate medical help seek medical attention.
- 3) **Inhalation:** Immediately remove the person to open area in fresh air. If person difficulty in breathing may seek medical attention.
- 4) **Eye Contact**: In case of eye contact with the material wash or flush eyes with fresh water, rinse eyes with warm water for 30 minutes and advised to seek medical attention.

SECTION 5 – FIRE FIGHTING MEASURES

(A)

- 1. Flammability: Hydrated Lime/Calcium Hydroxide is not considered as inflammable material
- 2. Flash Point: Not applicable to this material
- 3. Auto ignition Limit: Not Applicable
- 4. Upper & Lower explosive limits: Not Applicable
- 5. Explosion data: Not Applicable
- 6. Hazards Combustion products: None
- 7. Conditions to Avoid: None

(B) Methods:

Extinguishing medium In case of fire use water spray, dry chemical extinguisher, carbon dioxide or foam.

SECTION 6 – ACCIDENTAL RELEASE MEASURES

- 1. Avoid use of water on bulk spills. Person handling should have proper protective gears minimize generation of dust.
- 2. Avoid inhalation of Hydrated Lime and any contact with naked skin or eyes.
- 3. Do not wash huge quantity of Hydrated Lime in severs or water ways.
- 4. Residual amount can be flushed with large quantity of water.
- 5. Store spilled material in plastic or metal containers, avoid use of Aluminum.

SECTION 7 – HANDLING & STORAGE

Handling Procedures:

This material is mildly corrosive and reactive with water. Handle bags in a manner that will ensure minimal generation of dusts. Do not get on skin or do not breathe dust, which may generate accidentally. Follow safe work procedures and wear the appropriate personal protective equipment specified in Section 8. The workers must be instructed and trained in the safe work procedures.

Storage:

Keep dry, This product reacts with water to harden. Store in a sheltered area away from moisture. Do not store near foodstuffs. Store away from incompatible materials such as strong acids.

SECTION 8 – EXPOSURE CONTROLS / PERSONAL PROTECTION

Engineering Controls: Mechanical ventilation should be used to control dust levels.

Personal Protective Equipment

General: To identify additional Personal Protective Equipment (PPE) requirements, it is recommended that a hazard assessment in accordance with the OSHA PPE Standard (29CFR1910.132) or European Standard EN166 be conducted before using this product. 4.

Eye/ Face: Protection Use safety eyewear for protection against airborne particulate.

Skin Protection: Use impermeable clothing to prevent any contact with this product, such as gloves, apron, boots, or whole body suit. Recommended to prevent skin irritation in hypersensitive individuals. Natural rubber, neoprene, nitrile, or PVC gloves.

Respiratory Protection: In case of insufficient ventilation or extremely dusty environments, wear suitable respiratory equipment.

SECTION 9 – PHYSICAL AND CHEMICAL PROPERTIES						
Appearance : White or off-white powder	Odor : Odorless	Physical State : Solid				
Boiling Point (°C/°F): 2850 / 5162	Melting Point (°C/°F): 580 / 1076	Specific Gravity (Apparent) g/cc: 0.4 - 0.55 (True) g/cc: 2.2 - 2.4				
Vapor Pressure (mm Hg): N.A.	Vapor Density: N.A.	Evaporation Rate: N.A.				
Solubility in Water: Slightly soluble in water	pH (25°C/77°F): 12.4 approx					

SECTION 10 - STABILITY AND REACTIVITY

Stability: Stable but reacts slowly with carbon dioxide to form and calcium /magnesium carbonate. Keep dry until use. Hydrated lime may react with water, resulting in a slight release of heat, depending on the amount of lime (Calcium Oxide) present avoid contact with incompatible material.

Incompatibility: Wet Hydrated Lime and cement is alkaline and is incompatible with acids, ammonium salts and aluminum metal. Hydrated Lime and cement dissolves in hydrofluoric acid, producing corrosive silicon tetra fluoride gas. Hydrated lime and cement react with water to form silicates and calcium hydroxide. Silicates react with powerful oxidizers such as fluorine, boron tri fluoride, chlorine tri fluoride, manganese trifluoride and oxygen di-fluoride.

Hazardous polymerization: None.

Hazardous decomposition: Hydrated Lime will decompose at 540°C to produce calcium oxide (quicklime) magnesium oxide, and water.

SECTION 11 – TOXICOLOGICAL INFORMATION

Hydrated Lime is not listed by MSHA, OSHA, or IARC as a carcinogen, but this product may contain trace amounts of crystalline silica, which has been classified by IARC as carcinogenic to humans when inhaled in the form of quartz or crystobalite.

SECTION 12 – ECOLOGICAL INFORMATION

Environmental fate: Calcium Hydroxide or Slaked Lime is not expected to significantly bio accumulate.

SECTION 13 – DISPOSAL CONSIDERATIONS

Dispose of in accordance with all applicable federal, state, and local environmental regulations. If this product as supplied, and unmixed, becomes a waste, it will not meet the criteria of a hazardous waste as defined under statutory laws.

SECTION 14 – TRANSPORTATION INFORMATION.

Indian Transportation of Dangerous Goods Regulations:Not regulatedInternational Air Transport Association (IATA):Not regulated

International Maritime Organization (IMO): Not regulated

SECTION 15 – REGULATORY INFORMATION

EPA Regulations					
RCRA Hazardous Waste Number (40 CFR 261.33)	Not Listed				
RCRA Hazardous Waste Classification (40 CFR 261)	Not Listed				
CERCLA Hazardous Substance (40 CFR 261)	Not Listed				
CERLA Reportable Quantity (RQ)	Not Listed				
SARA 311/312 codes	Not Listed				
SARA Toxic Chemical (40 CFR 372.65)	Not Listed				
SARA EHS (Extremely Hazardous Substance) (40 CFR 355)	Not Listed				
Threshold Planning Quantity (TPQ)	Not Listed				
All components are listed on the USEPA TSCA Inventory List					
OSHA/MSHA Regulations					
Air Contaminant (29 CFR 1910.1000, Table Z-1)	Not Listed				
MSHA	Not Listed				
OSHA Specifically Regulated Substance (29 CFR 1910)	Not Listed				
State Regulations: consult state and local authorities for guidance					

SECTION 16 – OTHER INFORMATION

Prepared By: Synergy Chemical Industries, Jodhpur, India

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